



## Ultrafast intersystem crossing in naphthalene diimides: Involvement of higher triplet states

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Manuscript received online 30 April 2019, revised and accepted 16 May 2019

Conventionally, the triplet state generation via  $S_1 \rightarrow T_1$  is thought to be the slowest molecular photophysical process in organic conjugated molecules. However, recent advanced experimental spectroscopic investigations by Yushchenko *et al.* [*J. Phys. Chem. Lett.*, 2012, **6**, 2096-2100] revealed an ultrafast triplet formation of bromo core substituted naphthalene diimide (rNDI). Here, we estimated various stationary points of the low-lying singlet-triplet manifold of prototype NDIs using model vibronic Hamiltonian to gain insights into the mechanistic details of intersystem crossing (ISC) pathways of rNDI. Several ISC pathways involving  $S_1$  and  $S_2$  are identified where the higher triplet states,  $T_n$  ( $n > 2$ ) are found to be energetically more accessible than  $T_2$  and  $T_1$ . Our findings of NDIs are correlated to interpret the experimental ISC findings of rNDI.

Keywords: Intersystem crossing, ultrafast triplet formation, energy gap law, linear vibronic coupling, stationary points.

### Introduction

Internal conversion (IC) and intersystem crossing (ISC) are two competing nonradiative photophysical processes of a photoexcited molecule during its relaxation to the ground electronic state ( $S_0$ ). If the molecule is excited to Franck-Condon (FC) point on  $S_2$  state, then the molecule relaxes either to  $S_1$  state via IC or directly to an isoenergetic receiver triplet state via ISC. The other triplet-state formation pathway is the ISC from vibrationally relaxed  $S_1$  state. The latter process is the conventional ISC pathway referred in the literature<sup>1</sup>.

Recent experimental time-resolved nanosecond and femtosecond spectroscopic measurements revealed two different triplet generation timescales in amino and bromo core substituted red naphthalene diimide<sup>2</sup> (rNDI) (see Fig. 1). Fluorescence and triplet quantum yields were analyzed by exciting the molecule with  $S_2 \rightarrow S_0$  and  $S_1 \rightarrow S_0$  transition energies. The observed triplet formation timescale of  $<200$  fs was attributed to the direct ISC from  $S_2$  to triplet manifold. ISC involving  $S_1$  state is found to be slower, occurring on nanosecond time regime.

A detailed theoretical study to elucidate the reported experimental ISC pathways of rNDI is a daunting task. This is primarily due to the challenge posed by the energetically

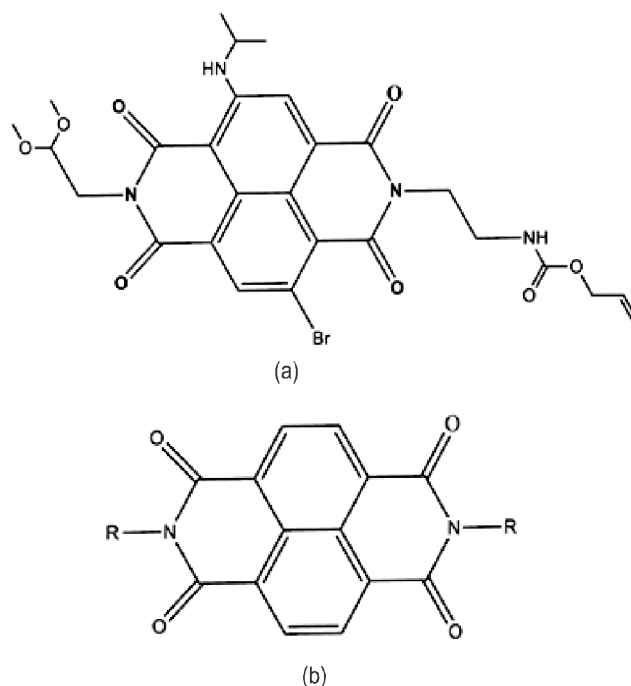


Fig. 1. Schematic diagram of (a) bromo and amino core substituted red naphthalenediimide, (rNDI) and (b) substituted NDI (R-NDI where R = H, Me).

close-lying excited electronic states, which tend to exhibit strong non-adiabatic effects. Moreover, the estimation of

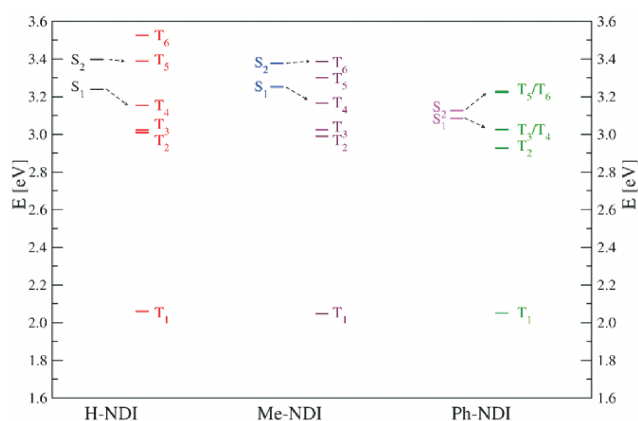
various stationary points of excited-state potential energy surfaces requires expensive quantum chemical calculations and can be very time consuming. Hence, we choose prototype NDI derivatives (see Fig. 1) to provide insight into ISC pathways of reported rNDI.

Two steps are involved in this study: First, vertical excitation energies are computed at FC geometry i.e.  $S_0$  equilibrium geometry, of each molecule to evaluate the singlet-triplet energy gaps. Such energy data is analyzed using energy gap law<sup>3,4</sup> to identify the energetically accessible receiver triplet states. Secondly, the location of global minima of low-lying singlet-triplet states with respect to FC point is the other factor which plays a key role in regulating ISC. In principle, excited-state optimization calculation could provide the minimum energy of the relevant state. However, such type of calculations are unfeasible owing to the complexity arising from the large number of vibrational degrees of freedom and near-degenerate electronic manifold, and are sometimes successful but not reliable. Hence, we employ a different strategy where simple model vibronic Hamiltonians within the well-established linear vibronic coupling<sup>5-7</sup> approach are constructed to estimate the global minima of the involved electronic states. The energy gap at FC geometry and at their respective minima of singlet-triplet manifold are taken into consideration for predicting the plausible ISC pathways in NDIs. Obtained outcomes are then correlated to interpret the experimental findings of rNDI.

## Results and discussion

**Geometry optimization:** Ground-state equilibrium geometry optimization and vibrational frequency calculations of NDIs are performed using density functional theory (DFT). A hybrid density functional method, B3LYP<sup>8</sup> (Becke, three parameter, Lee-Yang-Parr), in combination with 6-311G\* basis set is employed. Time-dependent DFT (TD-DFT) computations are carried out to evaluate the vertical excitation energies of singlet and triplet states of these molecules. All computations are carried out using Gaussian 09 software package<sup>9</sup>.

Fig. 2 depicts the vertical excitation energies of NDIs. From this Fig. 2, it is evident that  $T_1$  state is energetically well-separated from  $S_1$ , than the other triplet states, for all



**Fig. 2.** Vertical excitation energies of low-lying electronic states of R-NDI computed at TD-B3LYP/6-311G\* level of theory. Energetically favorable ISC pathways are shown with arrow symbol.

NDIs. Hence,  $S_1 \rightarrow T_1$  ISC pathway is less favourable, compared to other higher triplet receiver states, based on the “energy gap law” – which states that the rate of nonradiative transition should increase with decrease of energy gap of involved electronic states. We found that  $S_1 \rightarrow T_4$  in H- and Ph-NDI and  $S_1 \rightarrow T_5$  in Me-NDI are likely to be the energetically most favourable ISC pathways. If ISC originates from  $S_2$ , then  $S_2 \rightarrow T_6$  in H-NDI,  $S_2 \rightarrow T_5$  in Me-NDI and Ph-NDI would become the favourable pathways.

Note that the ISC pathway identified at FC point of  $S_2$  may compete with  $S_2 \rightarrow S_1$  IC. The latter process become dominant as the  $S_2$ - $S_1$  energy gap decreases with the increase of molecular size (*cf.* Fig. 2). The energy gap trends suggest that  $S_2 \rightarrow T_5/T_6$  ISC is favoured over  $S_2 \rightarrow S_1$  IC for H- and Me-NDIs. For Ph-NDI,  $S_2 \rightarrow S_1$  IC followed by ISC from  $S_1$  becomes the favourable triplet generation pathway.

Here, we highlight that the receiver triplet state in these favoured pathways may vary if the molecule undergoes very fast intramolecular relaxation to the respective minimum of involved singlet states due to severe geometrical distortions upon photoexcitation. The energy associated with these geometrical distortions is called the stabilization/reorganization energy of the electronic state ( $E_s$ )<sup>12</sup>. Details of geometrical distortions and  $E_s$  could be obtained by performing optimization calculation for an excited state. However, such a calculation is computationally expensive. Thus, as stated above,

we adopt an alternate procedure where the linear vibronic coupling mechanism is employed to evaluate various stationary points of excited-states.

A 2×2 (for  $S_1$ - $S_2$  states) and 6×6 (for  $T_1$ - $T_6$  states) model Hamiltonians are constructed within the linear vibronic coupling using diabatic representation and symmetry selection rules<sup>10</sup>. For instance, the Hamiltonian for  $S_1$ - $S_2$  coupled states is given as:

$$H = (T_N + V_0) I_2 + \begin{pmatrix} E_1 + \kappa_1 Q_g & \lambda Q_u \\ \lambda Q_u & E_2 + \kappa_2 Q_g \end{pmatrix}$$

where  $T_N$  and  $V_0$  represent the ground-state kinetic energy and potential energy terms.  $I_2$  represents 2×2 identity matrix. The potential energy term is approximated to be harmonic potential with  $V = S_i w_i Q_i^2$  where  $w$  is frequency and  $Q$  is the corresponding dimensionless normal coordinates<sup>11</sup> of vibration  $i$ . The totally and non-totally symmetric vibrational modes are represented by  $g$  and  $u$ , respectively.  $E_1$  and  $E_2$  are the FC vertical excitation energies of  $S_1$  and  $S_2$ , respectively. The interstate ( $k$ ) and intrastate ( $l$ ) coupling parameters are estimated using below expressions:

$$\kappa^{(n)} = \left( \frac{\partial V_n(Q)}{\partial Q} \right)_{Q_0}; n = 1, 2$$

$$\lambda^{(1,2)} = \left\{ \frac{1}{8} \frac{\partial^2}{\partial Q^2} \{V_1(Q) - V_2(Q)\}^2 \right\}_{Q_0}^{1/2}$$

A computational approach for evaluating  $k$  and  $l$  may be found elsewhere<sup>12</sup> and is not discussed here for brevity. In

the present study, we have followed this approach to estimate the parameters at TD-B3LYP/6-311G\* level of theory. The coupling constant values of NDIs are collected in Tables 1-3. The stabilization energy  $E_s$  is calculated using the expression:  $E_s = k^2/2w$ .

A quantum dynamical study of coupled  $S_1$ - $S_2$  and  $T_1$ - $T_6$  states is out of the scope of the present study and is planned for future investigations.

*Stationary points:* To understand on how the electronic energies vary with the geometrical distortions, we have plotted adiabatic potential energies, obtained by diagonalization of the diabatic Hamiltonian along the dimensionless normal coordinates of C=O stretching vibration of H-NDI in Fig. 3.

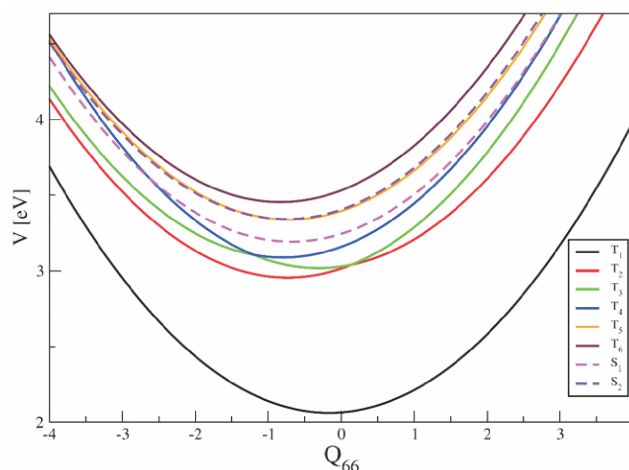


Fig. 3. Potential energy cuts of low-lying electronic states along C=O stretching mode ( $Q_{66}$ ) of H-NDI.

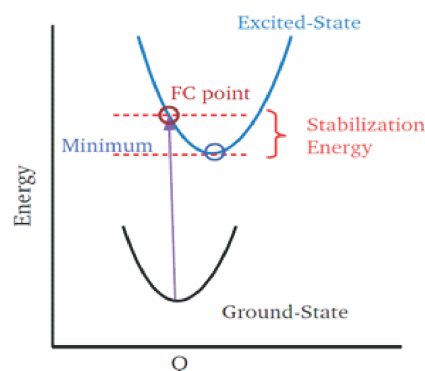
Table 1. Intrastate coupling parameters ( $k$ ) for excited singlet and triplet states for H-NDI. The stabilization energy ( $E_s$ ) corresponding to each vibrational mode is shown in the parentheses. All values are in eV

Mode (freq)	$S_1$	$S_2$	$T_1$	$T_2$	$T_3$	$T_4$	$T_5$	$T_6$
$\nu_{10}$ (0.0386)	-0.0183(0.1126)	-0.0168(0.0942)	-0.0271(0.0246)	-0.0178(0.1064)	0.0073(0.0181)	-0.0080(0.0215)	-0.0145(0.0704)	-0.0270(0.2449)
$\nu_{14}$ (0.0524)	0.0627(0.7170)	0.0388(0.2743)	0.0156(0.0445)	0.0578(0.6077)	-0.0086(0.0136)	0.0572(0.5958)	0.0365(0.2421)	0.0327(0.1946)
$\nu_{20}$ (0.0687)	0.0020(0.0004)	-0.0123(0.0160)	-0.0352(0.1312)	0.0028(0.0008)	-0.0193(0.0395)	0.0069(0.0050)	-0.0282(0.0840)	-0.0039(0.0016)
$\nu_{28}$ (0.0871)	-0.0251(0.0416)	-0.0248(0.0405)	-0.0463(0.1415)	-0.0222(0.0326)	0.0108(0.0077)	-0.0158(0.0164)	-0.0323(0.0688)	-0.0306(0.0617)
$\nu_{41}$ (0.1293)	-0.0418(0.0522)	-0.0460(0.0633)	-0.0566(0.0957)	-0.0440(0.0578)	-0.0909(0.2469)	-0.0510(0.0778)	-0.0508(0.0772)	-0.0694(0.1441)
$\nu_{44}$ (0.1418)	0.0533(0.0705)	0.0574(0.0820)	0.0333(0.0276)	0.0495(0.0608)	-0.0664(0.1095)	0.0528(0.0694)	0.0348(0.0302)	0.0355(0.0313)
$\nu_{50}$ (0.1659)	0.0686(0.0854)	0.0760(0.1048)	-0.0143(0.0037)	0.0639(0.0742)	0.0460(0.0384)	0.0692(0.0870)	0.0046(0.0004)	0.0150(0.0041)
$\nu_{53}$ (0.1743)	0.0097(0.0016)	0.0190(0.0059)	-0.1128(0.2093)	0.0108(0.0019)	-0.0057(0.0005)	0.0195(0.0062)	0.0359(0.0213)	-0.0672(0.0743)
$\nu_{54}$ (0.1775)	-0.1274(0.2575)	-0.1355(0.2913)	-0.1764(0.4940)	-0.1170(0.2171)	-0.1237(0.2428)	-0.1223(0.2373)	-0.0625(0.0620)	-0.0827(0.1086)
$\nu_{61}$ (0.2030)	0.1346(0.2197)	0.1510(0.2767)	0.2192(0.5830)	0.1299(0.2049)	0.0187(0.0042)	0.1239(0.1861)	0.2130(0.5506)	0.0700(0.0594)
$\nu_{68}$ (0.2211)	0.1507(0.2323)	0.1662(0.2824)	0.0364(0.0136)	0.1625(0.2702)	0.0688(0.0484)	0.1767(0.3194)	0.1573(0.2530)	0.1849(0.3497)
$\nu_{70}$ (0.3987)	-0.0015(0.0000)	-0.0070(0.0002)	-0.0101(0.0003)	-0.0010(0.0000)	-0.0097(0.0003)	-0.0062(0.0001)	-0.0049(0.0001)	-0.0072(0.0002)
$\nu_{72}$ (0.4448)	-0.0024(0.0000)	-0.0017(0.0000)	-0.0006(0.0000)	-0.0017(0.0000)	-0.0025(0.0000)	-0.0009(0.0000)	0.0006(0.0000)	0.0061(0.0001)

**Table 2.** Intrastate coupling parameters ( $k$ ) for excited singlet and triplet states for Me-NDI. The stabilization energy ( $E_s$ ) corresponding to each vibrational mode is shown in the parentheses. All values are in eV

Mode (freq)	$S_1$	$S_2$	$T_1$	$T_2$	$T_3$	$T_4$	$T_5$	$T_6$
$\nu_{12}$ (0.0365)	-0.0295(0.3259)	-0.0263(0.2598)	-0.0268(0.2701)	0.0006(0.0002)	-0.0282(0.2981)	-0.0210(0.1654)	-0.0227(0.1930)	-0.0112(0.0470)
$\nu_{14}$ (0.0388)	-0.0046(0.0070)	0.0025(0.0020)	0.0016(0.0008)	0.0001(0.0000)	-0.0045(0.0067)	-0.0048(0.0077)	0.0000(0.0000)	-0.0006(0.0001)
$\nu_{17}$ (0.0468)	0.0173(0.0686)	0.0008(0.0001)	0.0000(0.0000)	0.0005(0.0001)	0.0160(0.0586)	0.0176(0.0708)	0.0119(0.0324)	0.0105(0.0254)
$\nu_{19}$ (0.0509)	-0.0116(0.0259)	0.0012(0.0003)	0.0005(0.0000)	-0.0022(0.0009)	-0.0109(0.0231)	-0.0117(0.0263)	-0.0114(0.0249)	-0.0098(0.0185)
$\nu_{20}$ (0.0520)	0.0427(0.3372)	0.0002(0.0006)	0.0011(0.0002)	0.0059(0.0064)	0.0390(0.2812)	0.0424(0.3327)	0.0433(0.3466)	0.0381(0.2679)
$\nu_{26}$ (0.0688)	-0.0064(0.0044)	0.0330(0.1148)	0.0356(0.1336)	0.0215(0.0489)	-0.0068(0.0049)	-0.0110(0.0127)	0.0260(0.0715)	0.0261(0.0721)
$\nu_{29}$ (0.0767)	-0.0012(0.0001)	-0.0001(0.0000)	0.0005(0.0000)	0.0006(0.0000)	-0.0016(0.0002)	-0.0016(0.0002)	-0.0001(0.0000)	-0.0007(0.0000)
$\nu_{30}$ (0.0778)	0.0508(0.2135)	0.0298(0.0732)	0.0413(0.1407)	-0.0197(0.0319)	0.0458(0.1731)	0.0394(0.1283)	0.0071(0.0042)	-0.0007(0.0000)
$\nu_{39}$ (0.1064)	0.0001(0.0000)	-0.0002(0.0000)	0.0002(0.0000)	0.0006(0.0000)	0.0002(0.0000)	-0.0005(0.0000)	0.0030(0.0004)	0.0022(0.0002)
$\nu_{42}$ (0.1164)	0.0151(0.0085)	-0.0380(0.0532)	-0.0484(0.0866)	-0.0456(0.0767)	0.0117(0.0051)	0.0116(0.0050)	0.0292(0.0316)	0.0135(0.0068)
$\nu_{47}$ (0.1316)	0.0060(0.0010)	-0.0027(0.0002)	-0.0009(0.0000)	-0.0021(0.0001)	0.0057(0.0009)	0.0061(0.0011)	-0.0066(0.0013)	-0.0057(0.0009)
$\nu_{49}$ (0.1351)	0.0829(0.1883)	0.0329(0.0296)	0.0503(0.0693)	0.0490(0.0656)	0.0814(0.1814)	0.0893(0.2183)	0.0609(0.1017)	0.0535(0.0785)
$\nu_{52}$ (0.1457)	-0.0003(0.0000)	-0.0129(0.0039)	-0.0229(0.0124)	0.0780(0.1434)	0.0019(0.0001)	0.0039(0.0004)	-0.0110(0.0029)	0.0071(0.0012)
$\nu_{53}$ (0.1463)	0.0056(0.0007)	-0.0051(0.0006)	-0.0064(0.0009)	0.0218(0.0111)	0.0059(0.0008)	0.0058(0.0008)	-0.0036(0.0003)	0.0008(0.0000)
$\nu_{55}$ (0.1535)	0.0038(0.0003)	-0.0018(0.0001)	-0.0005(0.0000)	0.0020(0.0001)	0.0030(0.0002)	0.0025(0.0001)	0.0015(0.0001)	0.0015(0.0001)
$\nu_{58}$ (0.1619)	-0.0007(0.0000)	-0.0042(0.0003)	-0.0054(0.0006)	0.0019(0.0001)	-0.0007(0.0000)	0.0005(0.0000)	-0.0053(0.0005)	-0.0046(0.0004)
$\nu_{60}$ (0.1682)	0.0409(0.0295)	-0.0111(0.0022)	-0.0039(0.0003)	0.0390(0.0268)	0.0363(0.0233)	0.0391(0.0270)	0.0715(0.0905)	0.0701(0.0869)
$\nu_{63}$ (0.1741)	0.0245(0.0099)	-0.0451(0.0336)	-0.0553(0.0504)	0.0276(0.0125)	0.0228(0.0086)	0.0301(0.0150)	0.0706(0.0822)	0.0510(0.0429)
$\nu_{64}$ (0.1781)	0.1171(0.2160)	0.1662(0.4354)	0.2109(0.7008)	0.1179(0.2190)	0.1067(0.1794)	0.1104(0.1921)	0.1168(0.2152)	0.1199(0.2266)
$\nu_{66}$ (0.1816)	-0.0399(0.0242)	-0.0036(0.0002)	-0.0035(0.0002)	-0.0415(0.0262)	-0.0373(0.0211)	-0.0426(0.0275)	-0.0847(0.1088)	-0.0759(0.0874)
$\nu_{68}$ (0.1865)	0.0015(0.0000)	-0.0013(0.0000)	0.0002(0.0000)	0.0001(0.0000)	0.0007(0.0000)	0.0006(0.0000)	0.0001(0.0000)	-0.0003(0.0000)
$\nu_{72}$ (0.1885)	-0.0145(0.0029)	0.0049(0.0003)	0.0045(0.0003)	-0.0013(0.0000)	-0.0136(0.0026)	-0.0143(0.0029)	0.0003(0.0000)	0.0009(0.0000)
$\nu_{75}$ (0.2030)	0.1279(0.1986)	0.1674(0.3398)	0.2101(0.5357)	0.0268(0.0087)	0.1236(0.1855)	0.1189(0.1715)	0.1445(0.2532)	0.1292(0.2027)
$\nu_{76}$ (0.2052)	0.0036(0.0002)	0.0006(0.0000)	0.0042(0.0002)	0.0036(0.0002)	0.0035(0.0001)	0.0046(0.0002)	0.0024(0.0001)	0.0009(0.0000)
$\nu_{78}$ (0.2150)	0.0094(0.0010)	0.0098(0.0010)	-0.0001(0.0000)	-0.0010(0.0000)	0.0111(0.0013)	0.0119(0.0015)	-0.0052(0.0003)	-0.0017(0.0000)
$\nu_{80}$ (0.2195)	-0.1614(0.2704)	-0.0425(0.0187)	-0.0351(0.0127)	-0.0778(0.0628)	-0.1729(0.3101)	-0.1877(0.3655)	-0.1485(0.2288)	-0.1307(0.1772)
$\nu_{82}$ (0.3811)	-0.0023(0.0000)	-0.0011(0.0000)	-0.0007(0.0000)	-0.0017(0.0000)	-0.0029(0.0000)	-0.0034(0.0000)	-0.0066(0.0002)	-0.0064(0.0001)
$\nu_{86}$ (0.3964)	0.0039(0.0000)	0.0000(0.0000)	0.0000(0.0000)	-0.0005(0.0000)	0.0034(0.0000)	0.0039(0.0000)	-0.0043(0.0001)	-0.0039(0.0000)
$\nu_{88}$ (0.3974)	0.0001(0.0000)	0.0005(0.0000)	0.0004(0.0000)	0.0005(0.0000)	0.0001(0.0000)	0.0002(0.0000)	0.0001(0.0000)	0.0001(0.0000)
$\nu_{90}$ (0.3988)	0.0008(0.0000)	0.0085(0.0002)	0.0103(0.0003)	0.0090(0.0003)	0.0003(0.0000)	0.0052(0.0001)	0.0037(0.0000)	0.0049(0.0001)

The energy lowering of electronic states is evident from Fig. 3. For instance, the vertical excitation energy at FC geometry ( $Q = 0$ ) of  $S_1$  is  $\sim 3.24$  eV and the minimum on  $S_1$  potential curve is located at  $\sim 3.19$  eV (corresponding  $Q_{66} = -0.7$ ). The energy difference between these two stationary points is the  $E_s$  due to  $Q_{66}$ . For a better illustration,  $E_s$  is represented along with other stationary points in Fig. 4. Symmetry selection rules allows only the totally-symmetric vibrations would contribute to  $E_s$ . Total  $E_s$  ( $E_{st}$ ) is taken as an additive quantity, i.e. the contribution from each of the totally symmetric vibrational mode. Estimated  $E_{st}$  values of NDIs are given in Table 4.

**Fig. 4.** Stationary points of an excited state of a molecule.

**Table 3.** Intrastate coupling parameters ( $k$ ) for excited singlet and triplet states for Ph-NDI. The stabilization energy ( $E_{st}$ ) corresponding to each vibrational mode is shown in the parentheses. All values are in eV

Mode (freq)	$S_1$	$S_2$	$T_1$	$T_2$	$T_3$	$T_4$	$T_5$	$T_6$
$\nu_{12}$ (0.0219)	-0.0416(1.8010)	-0.0463(2.2380)	-0.0130(0.1749)	-0.0316(1.0412)	-0.0015(0.0023)	-0.0371(1.4311)	-0.0256(0.6807)	-0.0250(0.6521)
$\nu_{21}$ (0.0430)	0.0485(0.6351)	0.0519(0.7291)	-0.0130(0.1749)	-0.0316(1.0412)	-0.0015(0.0023)	-0.0371(1.4311)	-0.0256(0.6807)	-0.0250(0.6521)
$\nu_{33}$ (0.0645)	-0.0745(0.6679)	-0.0637(0.4871)	-0.0309(0.1151)	-0.0728(0.6378)	-0.0711(0.6071)	0.0103(0.0127)	-0.0775(0.7214)	-0.0811(0.7911)
$\nu_{35}$ (0.0679)	0.0031(0.0011)	-0.0137(0.0204)	-0.0293(0.0931)	0.0179(0.0348)	0.0111(0.0135)	-0.0223(0.0541)	-0.0377(0.1542)	-0.0380(0.1564)
$\nu_{42}$ (0.0824)	0.0090(0.0059)	-0.0011(0.0001)	0.0332(0.0811)	0.0180(0.0239)	0.0098(0.0070)	-0.0102(0.0077)	-0.0753(0.4174)	-0.0635(0.2967)
$\nu_{55}$ (0.1037)	0.0588(0.1609)	0.0723(0.2428)	-0.0357(0.0594)	0.0335(0.0523)	0.0501(0.1166)	-0.0154(0.0110)	0.0278(0.0360)	0.0279(0.0362)
$\nu_{71}$ (0.1267)	0.0205(0.0131)	0.0240(0.0180)	0.0080(0.0020)	0.0139(0.0060)	0.0185(0.0107)	0.0122(0.0046)	0.0259(0.0210)	0.0257(0.0206)
$\nu_{73}$ (0.1290)	0.0529(0.0841)	0.0567(0.0967)	0.0448(0.0604)	0.0449(0.0606)	0.0530(0.0845)	0.0651(0.1274)	0.0580(0.1012)	0.0579(0.1008)
$\nu_{75}$ (0.1308)	-0.0365(0.0390)	-0.0301(0.0264)	-0.0326(0.0311)	-0.0401(0.0470)	-0.0406(0.0483)	-0.0466(0.0634)	-0.0099(0.0029)	-0.0098(0.0028)
$\nu_{78}$ (0.1404)	0.0381(0.0368)	0.0289(0.0212)	0.0334(0.0282)	0.0478(0.0581)	-0.0370(0.0348)	0.0407(0.0420)	0.0206(0.0107)	0.0208(0.0109)
$\nu_{84}$ (0.1488)	-0.0179(0.0073)	-0.0244(0.0135)	-0.0029(0.0002)	-0.0035(0.0003)	-0.0120(0.0033)	0.0137(0.0042)	0.0575(0.0747)	0.0328(0.0242)
$\nu_{88}$ (0.1523)	0.0323(0.0225)	0.0217(0.0101)	-0.0161(0.0056)	0.0396(0.0339)	0.0372(0.0299)	0.0676(0.0986)	-0.0326(0.0229)	-0.0343(0.0254)
$\nu_{94}$ (0.1677)	-0.0355(0.0224)	-0.0523(0.0487)	0.0004(0.0000)	-0.0248(0.0110)	-0.0306(0.0166)	-0.0361(0.0232)	-0.0868(0.1340)	-0.0885(0.1393)
$\nu_{99}$ (0.1739)	0.0233(0.0090)	0.0364(0.0218)	-0.0531(0.0467)	0.0167(0.0046)	0.0224(0.0083)	0.0303(0.0152)	0.0538(0.0479)	0.0532(0.0469)
$\nu_{100}$ (0.1780)	-0.1256(0.2490)	-0.1361(0.2924)	-0.2096(0.6932)	-0.1098(0.1902)	-0.1213(0.2322)	-0.1178(0.2190)	-0.1342(0.2843)	-0.1347(0.2865)
$\nu_{106}$ (0.1903)	-0.0021(0.0001)	-0.0092(0.0012)	0.0043(0.0003)	0.0057(0.0004)	0.0007(0.0000)	0.0044(0.0003)	-0.0266(0.0097)	-0.0267(0.0098)
$\nu_{109}$ (0.2028)	0.1362(0.2254)	0.1400(0.2381)	0.2116(0.5446)	0.1260(0.1930)	0.1287(0.2013)	0.0185(0.0042)	0.1292(0.2031)	0.1294(0.2037)
$\nu_{113}$ (0.2043)	-0.0373(0.0167)	-0.0479(0.0275)	0.0178(0.0038)	0.0004(0.0000)	-0.0270(0.0088)	0.0087(0.0009)	0.0396(0.0188)	0.0719(0.0620)
$\nu_{118}$ (0.2206)	0.1599(0.2626)	0.1453(0.2170)	0.0374(0.0144)	0.1832(0.3447)	0.1850(0.3518)	0.0734(0.0554)	0.0945(0.0917)	0.0939(0.0905)
$\nu_{120}$ (0.3928)	0.0016(0.0000)	0.0025(0.0000)	0.0001(0.0000)	0.0004(0.0000)	0.0012(0.0000)	0.0004(0.0000)	0.0030(0.0000)	0.0029(0.0000)
$\nu_{124}$ (0.3951)	0.0029(0.0000)	0.0040(0.0001)	0.0000(0.0000)	0.0011(0.0000)	0.0023(0.0000)	0.0007(0.0000)	0.0040(0.0001)	0.0039(0.0000)
$\nu_{128}$ (0.3967)	-0.0133(0.0006)	-0.0192(0.0012)	0.0000(0.0000)	-0.0051(0.0001)	-0.0101(0.0003)	-0.0001(0.0000)	-0.0249(0.0020)	-0.0249(0.0020)
$\nu_{132}$ (0.3988)	-0.0029(0.0000)	-0.0040(0.0000)	-0.0103(0.0003)	-0.0012(0.0000)	-0.0040(0.0001)	-0.0095(0.0003)	-0.0043(0.0001)	-0.0043(0.0001)

**Table 4.** Total stabilization energy (in eV) for low lying excited states of R-NDI

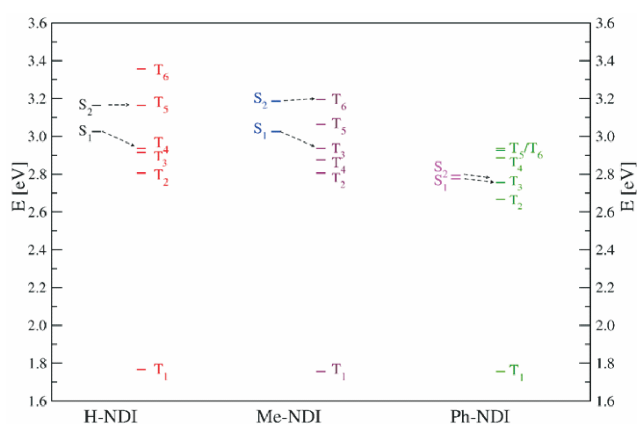
State	H-NDI	Me-NDI	Ph-NDI
$S_1$	0.22	0.23	0.31
$S_2$	0.23	0.19	0.33
$T_1$	0.30	0.30	0.30
$T_2$	0.21	0.11	0.26
$T_3$	0.11	0.22	0.27
$T_4$	0.22	0.23	0.14
$T_5$	0.22	0.24	0.30
$T_6$	0.17	0.19	0.29

$E_{st}$  of  $S_1$  is estimated to be  $\sim 0.22$ ,  $\sim 0.23$  and  $\sim 0.31$  eV for H-NDI, Me-NDI and Ph-NDI, respectively. We found that the major contribution ( $\sim 20\%$ ) to this energy is originating from C=O symmetric stretch ( $\nu_{66}$ ) vibrational mode.

Also, similar  $E_{st}$  values found for  $S_2$  of NDIs. These  $E_{st}$  values suggest that the geometrical distortions in  $S_1$  and  $S_2$  are not severe, as is expected due to their rigid polycyclic skeletal structure.

Finally, the global minimum of a given electronic state is estimated by subtracting the  $E_{st}$  from FC vertical energy. Estimated global minima of electronic states of NDIs are plotted in Fig. 5.

Energy gap trends of electronic state minima (*cf.* Fig. 5), suggest that  $T_4$  and  $T_5$  to be the receiver triplet states, same as at FC geometry, for ISC occurring via  $S_1$  and  $S_2$  states, respectively, in H-NDI. On the other hand,  $T_5$  in Me-NDI and  $T_3$  in Ph-NDI to be the receiver triplet state for ISC via  $S_1$ . Similarly,  $T_6$  in Me-NDI and  $T_3$  in Ph-NDI are found more relevant for ISC via  $S_2$ .

**Fig. 5.** Equilibrium minimum energies of low-lying electronic states of R-NDI computed at TD-B3LYP/6-311G\* level of theory. Energetically favorable ISC pathways are shown with arrow symbol.

In summary, high-lying triplet states ( $T_3$ - $T_6$ ) are responsible for ISC originating either at respective FC point or at equilibrium minimum of  $S_1$  and  $S_2$  of R-NDIs. All these involved states lie  $\sim 4000\text{ cm}^{-1}$  above  $T_2$ , indicating a minimal or no direct participation of  $T_2$  (and  $T_1$ ) in ISC of NDIs.

Assuming the energetics and associated ISC pathways of rNDI to be same as that of R-NDI, then an enhancement of ISC efficiency is thus expected for rNDI due to heavy-atom effect<sup>13</sup>. Furthermore, presence of Br atom and core-substitution may induce additional electronic effects thereby resulting in various unexpected ISC pathways in rNDI. Future studies are aimed in this direction to provide an accurate understanding of photophysical properties of rNDI.

### Conclusions

Energetics of involved singlet-triplet manifold play a key role in molecules exhibiting ultrafast intersystem crossing process in the absence of a heavy atom. Our findings on low-lying singlet-triplet manifold clearly indicate the minimal or no direct participation of  $T_1$  and  $T_2$  states in ISC in NDIs. Higher receiver triplet states ( $T_n$ ,  $n > 2$ ), isoenergetic with  $S_1$  and  $S_2$ , have to be considered for interpreting the mechanistic details of ISC in NDIs.

The insights obtained from our study may pave the way for a better understanding of photophysical processes of NDIs<sup>14-16</sup> and trigger new experimental and theoretical studies to control and design optoelectronic materials<sup>17,18</sup>.

### Acknowledgements

This project is funded by Science and Engineering Research Board (SERB), Department of Science and Technology (DST), Government of India (File No. ECR/2016/000226). SVK acknowledges DST for INSPIRE-PhD fellowship.

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